

Calea Mănăștur 3-5, 400372, Cluj-Napoca Tel: 0264-596.384, Fax: 0264-593.792

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No.	1	
INO.	/	

Form code USAMV 0701020114

SUBJECT OUTLINE

1. General data

1.1. Higher education institution	University of Agricultural Sciences and Veterinary Medicine Cluj-Napoca
1.2. Faculty	Food Science and Technology
1.3. Department	Food Sciences
1.4. Domain of study	Food Engineering
1.5. Level of study ¹⁾	Bachelor of Science
1.6. Specialization	Technology of agricultural products processing
1.7. Form of education	IF

2. Date despre disciplină

2.1. Name of the cours	se	Food biochemistry						
2.2. Course coordinator Lecturer.dr. Zorita Diaconeasa								
2.3. Coordinator of the laboratory/seminar activity			Lecturert.	dr. Zorita Dia	coneasa			
2.4 Voor of study	TT	2.5. Semester	1	2.6. Type of	Continuous	2.7. Course	Content ²	DF
2.4. Year of study	11	2.3. Semester	4	evaluation	Summative	regime	Compulsory level ³	DI

3. Total estimated time (hours/semester for the teaching activities)

3.1. Number of hours/week-full time form	4	out of which: 3.2. course	2	3.3. seminar/ laboratory/ project	2
3.4. Total hours in the teaching curricula	56	out of which: 3.5. course	28	3.6. seminar/laboratory	28
Distribution of time					hrs.
3.4.1. Study based on hand book, notes, bibliography				5	
3.4.2. Extra documentation in library, on specific electronic platforms and on field				5	
3.4.3. Preparation of seminars / laboratories / projects, essays, reports, portfolios				5	
3.4.4. Tutorial				2	
3.4.5. Examination				2	
3.4.6. Other activities					
3.7. Total hours of individual study	19				1

3.7. Total hours of individual study	19
3.8. Total hours on semester	75
3.9. Number of credits ⁴	3

4. Pre-conditions (where relevant)

4.1. for curriculum	Organic Chemistry
4.2. for competences	The student must have knowledge regarding general chemisty and organic chemistry from
	highschool order to understand important food chemistry principles

5. Conditions (where relevant)

5.1. for course	The course is interactive; students can ask questions regarding the content of the		
	exposure. Academic discipline enforces time start and end of the course. Any other		
	activities are forbidden during course, the mobile phones must be switched off.		
5.2. for seminar/laboratory/project	In the laboratory students must consult the practical guide, every student will		
	perform individual activity using the laboratory equipment which is described in		
	the practical guide. During the practical activities the academic discipline must be		
	maintainted.		

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6. Specific competences acquired

Competențe profesionale	-Description and use of concepts, theories and basic methods used in quality control of food products; the concepts are refering to the chemical compounds that assure the product quality, their transformation during processing, transportation and storage, the equipment and the quantification methods used for determining these compounds -Description and use of concepts, theories and methods of basic Food science (defined in multidisciplinary terms), on the structure, properties and transformations of food compounds and contaminants throughout the food chain -Explanation and interpretation of concepts, processes, models and methods of food science, using basic knowledge on the composition, structure, properties and transformations of food compounds and their interaction with other systems throughout the food chain
Competențe transversale	- Applying strategies like perseverance, rigor, efficiency and responsibility in work, punctuality and personal assumption of responsibility for business results, creativity, common sense, analytical and critical thinking, problem solving and so on, based on principles, norms and code values applied for

7. Objectives of the course (as a result of the specific competences acquired)

7.1 General objectives	Assimilation of fundamental concepts of biochemistry required for engineers in the food industry in order to understand and learn other			
	disciplines (nutrition, toxicology, food control, etc.); knowledge of			
	organic compounds involved in the proper functioning of plant and			
	animal organisms.			
7.2. Specific objectives	Food Biochemistry course aims to prepare sudents for senior years, help			
	them to achieve an understanding of the chemical changes that take			
	place with food components during processing and storage.			
	The study of biochemistry is necessary for arming students with the			
	knowledge and practical skills on the handling of laboratory tools,			
	identification or determination of chemical compounds based on its			
	content.			

8. Content

8.1. COURSE	Teaching methods	Observation
Number of hours – 14 1. The chemical composition of vegetable and animal organisms: inorganic and organic compounds founds in food. Bioelements. Hierarchy of molecular organization of organisms. Biomolecule relationship - cell structure	Lectures	1 Lecture
2.Carbohydrate metabolism: carbohydrates anabolism; Photosynthesis and chemosynthesis; Oligoglucidelor and poliglucidelor biosynthesis. Carbohydrate catabolism: Glycolysis. Krebs's cycle. Oxidative phosphorylation and tissue respiration; Pentosophosphates cycle and fermentative degradations; Vitamin C biosynthesis.	Lectures	1 Lecture
3.Lipid metabolism: The role of lipids in plant organism. Glycerol metabolism. The biosynthesis of saturated and unsaturated fatty acid. The biosynthesis of triglycerides. Glycerides Catabolism:	Lectures	1 Lecture



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degradation of saturated and unsaturated fatty acids through β-oxidation and α-oxidation. Biosynthesis and degradation of phosphatides and sfingoglicolipid		
4. Protein metabolism: The role of proteins in the body. Amino acids anabolism; General methods for the biosynthesis of amino acids (reductive amination IVb) Amino acids catabolism. Degradation by deamination and decarboxylation. Ammonia metabolism. Ureogenetic cycle and glutamic acid cycle. Biosynthesis and degradation of nitrogenous bases. Biosynthesis and degradation of DNA and RNAs. The mechanism of proteins biosynthesis.	Lectures	1 Lecture
5. Biochemistry degradation processes: Influence of light on the chemical composition of food (degradation of proteins, fats, pigments, vitamins).	Lectures	1 Lecture
6. Changes occurring in food as a result of heat treatment: the thermal degradation of sugars, proteins, lipids, vitamins	Lectures	1 Lecture
7. Non-enzymatic oxidative degradation of foodstuffs: Processes of oxidative rancidity of fat, of carotenoids, tanning substances, substances der flavor. Prevention of autoxidation processes.	Lectures	1 Lecture
8. Food degradation under the action of heavy metals and polluted them with pesticides	Lectures	1 Lecture
9. Food spoilage caused by enzymes and prevent these processes.	Lectures	1 Lecture
10. Biochemistry preservation processes: Freezing food; Biochemical transformations of proteins, lipids, vitamins freezing process. The enzymatic activity in congelate. Transformări products occurring in the milk, fish, meat, eggs, dairy products during freezing and storage. Conservation by lactic fermentation and antiseptic food.	Lectures	1 Lecture
11. Changes and degradation of dehydrated food: Correlations between water activity and lipid oxidation, browning carbohydrates, enzyme reactions. Biochemical changes occurring in plant products, milk, dairy products and dehydrated meat.	Lectures	1 Lecture
12. Conservation through chemical sterilization: Influence of thermal sterilization of canned vegetables, the meat and milk.	Lectures	1 Lecture
13. Special Biochemistry: biochemical transformations during storage of fruits and vegetables - carbohydrates transformations; their pectin substances, organic acids, carotenoids, vitamins.	Lectures	1 Lecture
14. Special Biochemistry: The coagulation of milk, cheese ripening, maturation of meat, milk and crops biochemical processes in fermented dairy products.	Lectures	1 Lecture



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8.2. PRACTICAL WORK Number of hours – 28		
Quantification of carbohydrates in food by HPLC; The pectin from fruits used for jellies obtaining.	Practical work	2 Practical work
2 . Lipids extraction from chocolate, chips and sunflower seeds.	Practical work	2 Practical work
3. Determination of the main constants of glycerides: saponification, iodine, acid value, peroxide index.	Practical work	3. Practical work
4. The separation of a mixture of amino acids by paper chromatography. Quantification of meat proteins by spectrophotometric method.	Practical work	2 Practical work
5. Evaluation of enzymatic activity depending on temperature and pH.	Practical work	2 Practical work
6. Determination of biochemical changes in plant pigments by heat treatment; Biochemistry of pigments in meat	Practical work	2 Practical work
7. Practical examination	Exam	Practical examination

Compulsory bibliography:

- 1. G. Neamţu "Biochimie Alimentara" Edit. Ceres, Bucureşti, 1997
- 2. Andreea Stănilă, Carmen Socaciu, "Biochimia alimentelor- Lucrări practice și teste", Editura Academic Press, Cluj-Napoca, 2004
- 3. Andreea Stănilă Analiza compusilor bioactivi din alimente; Ed.Academic Press Cluj-Napoca; 2013 -

Facultative bibliography:

- 1) L.Stryer "Biochemistry"-fourth edition, W.H.Freeman & Co., New York, 1995
- 2) Doina Miere Chimia si igiena alimentelor; Ed. Medicală Universitara Iuliu Hatieganu, Cluj-Napoca, 2002

9. Corroboration of the subject content with the expectations of the epistemic community representatives, of the professional associations and representative employers in the domain

In order to identify ways of modernization and continuous improvement of teaching and course content, with the current issues and practical problems, teachers attend the annual meeting of the Association of Specialists in Food Industry of Romania as well as business meetings with members of food industry.

10. Evaluation

Type of activity	10.1. Evaluation criteria	10.2. Evaluation methods	10.3. Percent of the final grade
10.4. Course	Identify the main classes of organic	Oral	
	compounds.	Exam	75%
	Knowledge of organic chemical		
	reactions, identification of		
	mechanisms reaction.		
	Knowing the properties of classes of		



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	organic compounds found in the food		
	industry		
10.5. Seminar/Laboratory	Theoretical and practical knowledge	periodic evaluation / colloquy	25%
	of the methods of analysis used in the		
	chemistry lab.		
	Troubleshooting stoichiometric with		
	practical applicability (concentration,		
	purity, yield).		
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10.6. Minimal standard of performance

Mastering scientific information conveyed through lectures and practical work at an acceptable level. Obtaining the pass mark in continuous assessment is the condition of graduation.

Level of study- to be chosen one of the following - Bachelor/Post graduate/Doctoral

² Course regime (content) – for bachelor level it will be chosen one of the following - **DF** (fundamental subject), **DD** (subject in the domain), **DS** (specific subject), **DC** (complementary subject).

³ Course regime (compulsory level) - to be chosen one of the following - **DI** (compulsory subject), **DO** (optional subject), **DFac** (facultative subject)

One ECTS is equivalent with 25-30 de hours of study (didactical and individual study).

Filled in on 8.09.2021

Course coordinator Lecturer.dr. Zorita Diaconeasa Laboratory work/seminar coordinator Lecturer.dr. Zorita Diaconeasa

Subject coordinator Lecturer.dr. Zorita Diaconeasa

Head of the Department Prof. Ramona Suharoschi, PhD

Dean Prof. Elena Mudura, PhD

Approved by the Faculty

Approved by the

Department on 22.09.2021

Council on 28.09.2021